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### ANALYTICAL METHOD DEVELOPMENT AND VALIDATION FOR THE ESTIMATION OF ETELCALCETIDE IN BULK AND ITS DOSAGE FORM USING RP-HPLC

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#### ABSTRACT

A simple and precise RP-HPLC method was developed for the estimation of the Etelcalcetide. Initially, various mobile phase compositions were tried to separate title ingredients. Mobile phase and flow rate selection was based on peak parameters (height, tailing, theoretical plates capacity or symmetry factor) run time and resolution. Finally The mobile phase containing mixture of 50% OPA buffer 50% methanol with the flow rate of 1.0 ml/ min. The optimum wavelength for detection was 235nm at which better detector response for drug was obtained. The Retention time for Etelcalcetide was found to be 4.653min respectively. To ascertain its effectiveness, system suitability test were carried out on fresh prepared stock solutions. The regression 0.999 respectively. The low value of % R.S.D indicate the method is precise and accurate.

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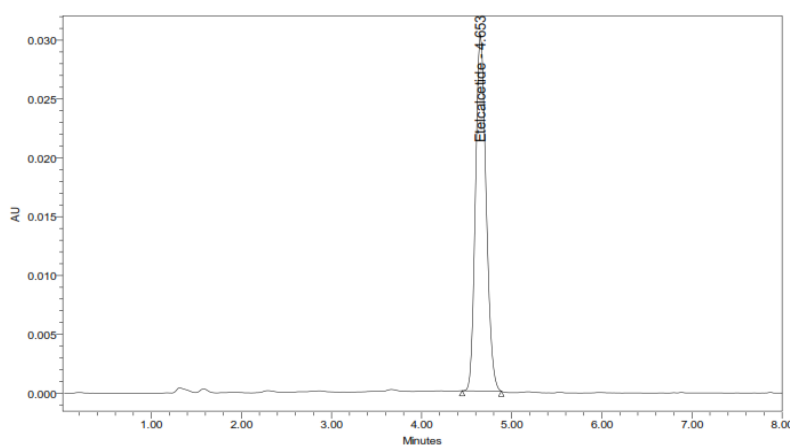
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of system suitability parameters were shown in table 1. The chromatogram of working standard solution is shown in fig 1. The summary of chromatographic condition was in table.

**Table 1: Summary of Chromatographic conditions.**

| S. No | Parameter            | Description/Value                                     |
|-------|----------------------|-------------------------------------------------------|
| 1.    | Stationary Phase     | Inertsil ODS C <sub>18</sub> Column (4.6mm x15mm)5µm. |
| 2     | Mobile Phase         | 0.1% Orth phosphoric acid buffer, methanol (50:50)    |
| 3     | Flow rate            | 1ml/min                                               |
| 4     | Detection Wavelength | 250 nm                                                |
| 5     | Detector             | Photo diode array                                     |
| 6     | Injection            | Auto sampler -Waters, model 717 plus                  |
| 7     | RT                   | 4.653Min                                              |
| 8     | Injection volume     | 20 µl                                                 |
| 9     | Column Temperature   | 35 °C                                                 |
| 10    | Run time             | 8 mins                                                |
| 11    | Diluent              | Mobile Phase                                          |
| 12    | Mode of separation   | Isocratic mode                                        |



**Fig.3 Typical Chromatogram of Etelcalcetide.**

**Table 2: System suitability parameters.**

| S. No | Parameter              | Result               |
|-------|------------------------|----------------------|
|       |                        | <b>Etelcalcetide</b> |
| 1     | Retention Time         | 4.653min             |
| 2     | Tailing                | 1.18                 |
| 3     | Theoretical Plates (n) | 3828                 |
| 4     | Similarity Factor      | 1.25                 |

#### Method Validation:

##### Accuracy

Recovery assessment was obtained by using standard addition technique which was by adding known quantities of pure standards at three different levels in 50%, 100% and 150% to the pre analyzed sample formulation. From the amount of drug found, amount of drug recovered and percentage recovery were calculated which sense to conformation that the proposed method was accurate. The results were tabulated in Table 3.

**Table 3: Results of Accuracy of Etelcalcetide.**

| %Concentration<br>(at specification Level) | Area   | Amount Added<br>(mg) | Amount Found<br>(mg) | % Recovery | Mean Recovery |
|--------------------------------------------|--------|----------------------|----------------------|------------|---------------|
| 50%                                        | 95505  | 50                   | 49.74                | 99.47*     | 99.40**       |
| 100%                                       | 191399 | 100                  | 99.67                | 99.67*     |               |
| 150%                                       | 285309 | 150                  | 148.58               | 99.05*     |               |

\*Mean % Recovery of 6 replicates; \*\*Mean % Recovery of 3 replicates.

### Precision

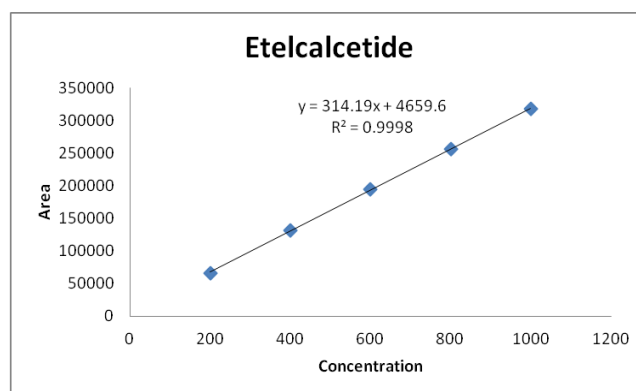
The intraday and interday precision of the proposed method was determined by analyzing Etelcalcetide, 3 times on the same day and on 3 different days. The results shown in table 4 were reported in terms of relative standard deviation.

**Table 4: Results of Precision (% Assay).**

| Injection     | Area     |
|---------------|----------|
| 1             | 191345   |
| 2             | 191232   |
| 3             | 191671   |
| 4             | 191999   |
| 5             | 192898   |
| 6             | 194679   |
| Average Assay | 192304.0 |
| STD           | 1308.1   |
| % RSD         | 0.7      |

### Linearity

Calibration graphs were constructed by plotting peak area vs concentration of Etelcalcetide and regression equations were calculated. The calibration graphs were plotted and over 5 different linear concentrations in the range of 5.



**Figure 4: Calibration graph for Etelcalcetide**

### Limit of detection (LOD) and limit of quantitation (LOQ):

The limit of detection (LOD) and limit of quantitation (LOQ) were determined by calculating the signal-to-noise(S/N) ratio 3.00 $\mu$ g/ml. Respectively according to International conference on Harmonization guidelines LOD value for Etelcalcetide were found to be 9.98 $\mu$ g/ml.

### Assay of the tablet dosage form

The proposed validated method was successfully applied to determine etelcalcetide in dosage form. The results obtained is comparable with corresponding label amount. The results were tabulated in the table.

### DEGRADATION STUDIES:

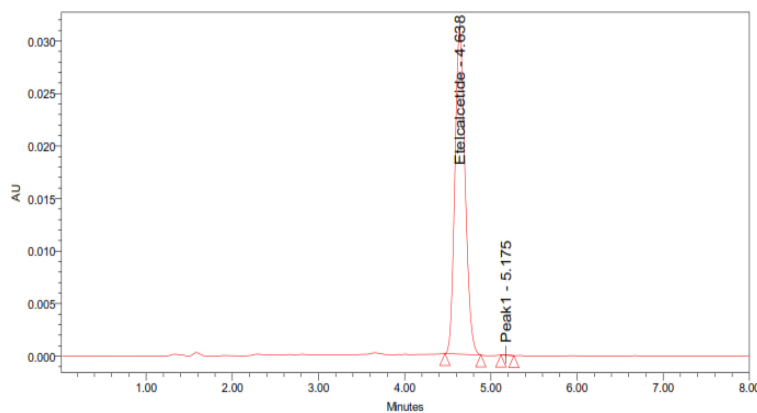
As per the ICH guidelines force degradation studies are conducted to show the degradation mechanisms under the various stress conditions like hydrolysis (i.e., acid, base, and water), oxidation, heat, and photolytic degradation was carried out. As well as these are simple, precise, economic and rapid to perform.

### Preparation of stock:

Accurately Transferred equivalent solution of 100 mg of Etelcalcetide into a 10 ml dry VF added few ml of Diluent and made to dissolved completely and volume is made with a diluent. Then is Filtered through 0.44 micron Injection filter. (Stock solution)

### Hydrolytic degradation under acidic condition

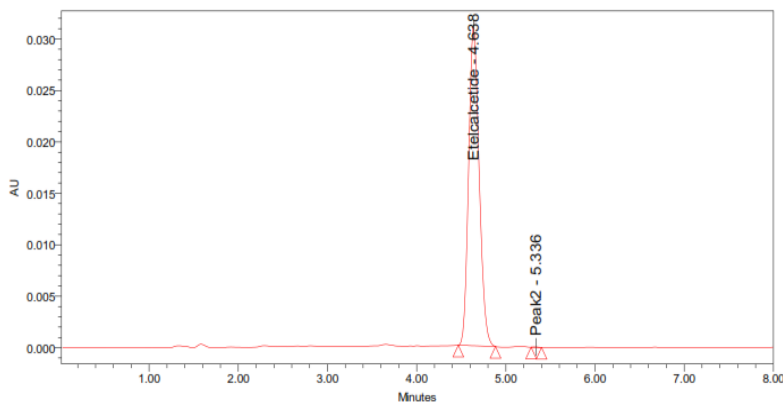
Pipette 0.6 ml of above solution into a 10ml volumetric flask and 3 ml of 0.1N HCl was added. Then, the VF was kept at 60°C for 24 hours and then neutralized with 0.1 N NaOH and make up to 10ml with diluent.



**Figure 5: Chromatogram showing Acid degradation.**

#### Hydrolytic degradation under alkaline condition

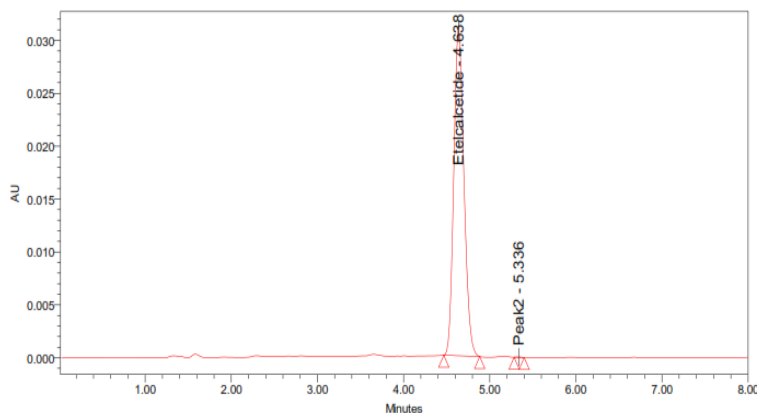
Pipette 0.6 ml of above solution into a 10ml volumetric and add 3ml of 0.1N NaOH was added in 10ml of volumetric flask. Then, the volumetric flask was kept at 60°C for 24 hours and then neutralized with 0.1N HCl and make up to 10ml with diluent.



**Figure 6: Chromatogram showing Alkaline degradation.**

#### Thermal induced degradation

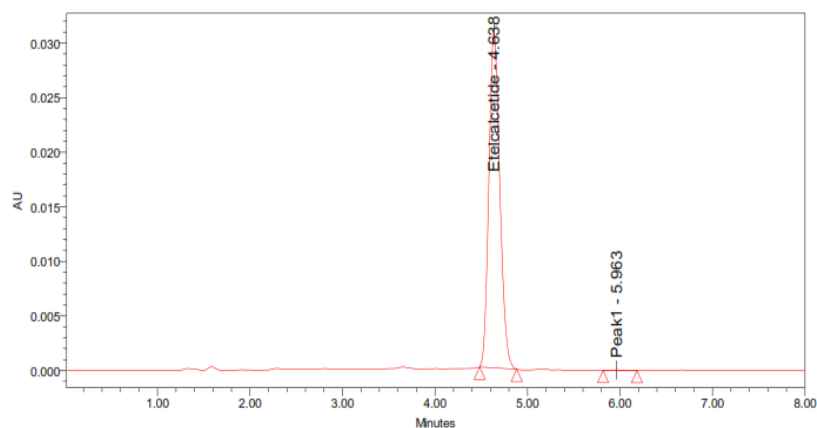
Etelcalcetide sample was taken in Petri dish and kept in Hot air oven at 110°C for 3 hours. Then the sample was taken and diluted with diluents and injected into HPLC and analyzed.



**Figure 7: Chromatogram showing Thermal degradation.**

#### Oxidative degradation

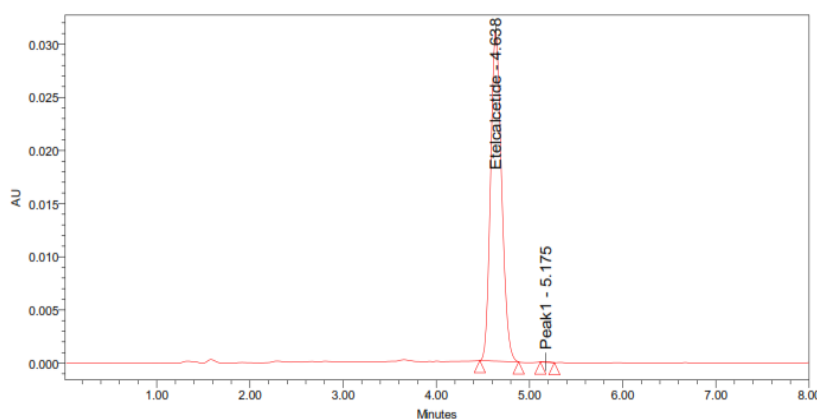
Pipette 0.6 ml above stock solution into a 10ml volumetric flask and 1ml of 30% w/v of hydrogen peroxide added in 10 ml of volumetric flask and the volume was made up to the mark with diluent.



**Figure 8: Chromatogram showing Peroxide degradation.**

#### Photo degradation

Pipette 0.6 ml above stock solution into a 10ml volumetric flask and exposed to sunlight for 24hrs and the volume was made up to the mark with diluents



**Figure 9: Chromatogram showing Photo degradation.**

**Table 05: Degradation results for Etelcalcetide.**

| Sample Name | Etelcalcetide |            |
|-------------|---------------|------------|
|             | Area          | % Degraded |
| Standard    | 191642        |            |
| Acid        | 183252        | 4.38       |
| Base        | 183532        | 4.23       |
| Peroxide    | 183253        | 4.38       |
| Thermal     | 187552        | 2.13       |
| Photo       | 186452        | 2.71       |

#### CONCLUSION

The proposed method has advantage of simplicity and convenience for the separation and quantitation of Etelcalcetide which can be used for the assay of their dosage form. Also, the low solvent consumption and short analytical run time lead to environmentally friendly chromatographic procedure. The method is accurate, precise, rapid and selective for estimation of Etelcalcetide. Hence it can be conveniently adopted.

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**ABBREVIATIONS:**

|         |                                                                 |
|---------|-----------------------------------------------------------------|
| RP-HPLC | : Reverse phase high performance/pressure liquid chromatography |
| CKD     | : Chronic kidney disease                                        |
| VF      | : Volumetric flask                                              |
| LOD     | : limit of detection                                            |
| LOQ     | : limit of quantitation                                         |
| OPA     | : Ortho phosphoric acid                                         |

**CONFLICT OF INTEREST:**

There is no conflict of interest.

**REFERENCES**

1. CBS publications and distributors, 2005.
2. P.D. Sethi, HPLC quantitative analysis of pharmaceutical formulations CBS publications and distributors, 1<sup>st</sup> edition, 2001.
3. B.K Sharma, instrumental method of chemical analysis, 23<sup>rd</sup> edition, goal publishers 2004.
4. Practical HPLC method development Lloyd R.Snyder, Joseph J. Kirkland, Joseph L. Glajch, second edition.
5. Validating chromatographic methods, David M.Bliesner.
6. International conference on harmonization: ICH Q 2 (R1) Validation of Analytical Procedures: Text and Methodology 1995.
7. R. Meyer Veronica, Practical High Performance Liquid Chromatography, 2nd Ed (1993) 26, 27, 40, 222, 246 and 258.
8. [http://hplc.chem.shu.edu/NEW/HPLC\\_Book/](http://hplc.chem.shu.edu/NEW/HPLC_Book/)
9. Szepei Gabor., HPLC in pharmaceutical Analysis, Vol. I (1990) 101-173.
10. G. H. Jeffery, J. Bassett, Vogel's textbook of Quantitative Chemical Analysis, 5th Ed (1991) 217-235.
11. H. Willard Hobart, L. L. Merritt, A. Dean John, Instrumental Methods of Analysis, 7th Ed, CBS Publishers, 580-610.
12. B.K. Sharma, Instrumental Methods of Chemical Analysis, 20th Ed (2001) 54-83.
13. S. Naga Sindhu, Y Srinivasa Rao, T. Hemant Kumar and K. Vera Prasada Rao, Method development and validation of RP-HPLC method for estimation of imatinib mesylate in pure and pharmaceutical dosage form, Der Pharmacia Lettre, 2015, 7 (3):33-38.
14. P. Sandhya, P.Vishnu Priya, Shyamala, N. Anjali Devi and JVC. Sharma, Method Development And Validation Of Imatinib Mesylate In Pharmaceutical Dosage Form By RP-HPLC, World Journal of Pharmacy and Pharmaceutical Sciences, Volume 3, Issue 1, 682-688.
15. Suresha D N, Pramila T And Tamizh Mani T, Method Development And Validation Of Imatinib Mesylate – Review, International Journal Of Pharmacy And Pharmaceutical Analysis (2016, Vol. 01(03), pg. 1-11.



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