



## A PHAMACOGNOSTICAL APPROACH IN ISOLATION OF BOSWELLIC ACIDS FROM BOSWELLIA SERRATE

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### ABSTRACT

**Olibanum** also known as “Dhup”, Indian Frankincense is an oleo gum resin of *Boswellia species*. In India it is obtained from *Boswellia serrata*. *Boswellia serrata* (Burseraceae), much branched, deciduous tree that grows abundantly in the dry, hilly parts of India. In India, the gum resin exudates of *Boswellia serrata*, known in the vernacular as “Salai guggal”, has been used in the Ayurvedic system of medicine in the managements of several inflammatory conditions and as a topical anti-inflammatory agent. The major use of *Boswellia serrata* in contemporary medicine is as an anti-arthritic and anti-inflammatory pharmacological agent. The anti-inflammatory properties of the gum

resin are attributed to the presence of “boswellic acids”. In 1992, the active principles within the multi-component mixture of resin were identified, resulting in recognition of Boswellic acids. The most important are Acetyl 11-Keto  $\beta$ -Boswellic Acid (AKBA) and 11- Keto  $\beta$ -Boswellic Acid (KBA). **Boswellic acids were found to inhibit two pro-inflammatory enzymes**, 5-lipoxygenase (which generates inflammatory leukotrienes) and Human Leukocyte Elastase (HLE). HLE is a serine protease that initiates injury to the tissue, which in turn triggers the inflammatory process. This dual inhibitory action on the inflammatory process is unique to boswellic acids. *Boswellia* species are trees or shrubs with an outer bark often peeling in parchment like flakes, inner bark greenish, with watery aromatic resin, wood with milky latex. The leaves are imparipinnate, mostly congested at the end of the branches. Flowers are bisexual in panicles or racemes. *Boswellia* trees are found at areas from the sea level up to 1000 meters, usually in rocky slopes and gullies, often on limestone boulders, more rarely on vertical rock-faces, growing to a height of 3 up to 12 meters. The four major pentacyclic triterpenic acids present in the acidic extract of *Boswellia serrata* gum resin.

- $\beta$ -Boswellic Acid
- Acetyl- $\beta$ -Boswellic Acid
- 11-keto- $\beta$ -Boswellic Acid
- Acetyl-11-keto- $\beta$ -Boswellic Acid

Drug occurs in globular, transparent, tears forming agglomerates of various shapes and sizes, brownish-yellow, upto 5 cm long, 2 cm thick, fragrant, fracture brittle; fractured surface waxy and translucent; burns readily and emanates an agreeable characteristic, balsamic resinous odor; taste, aromatic and agreeable.

**KEYWORDS:-** Boswellic acid, olibanum, Human Leukocyte Elastase, 11- Keto  $\beta$ -Boswellic Acid, anti-inflammatory, hilly parts of India, inflammatory leukotrienes, Burseraceae.

### INTRODUCTION To MATERIAL

**Olibanum** also known as “Dhup”, Indian Frankincense is an oleo gum resin of *Boswellia species*. In India it is obtained from *Boswellia serrata*. *Boswellia serrata* (Burseraceae) is a large, much branched, deciduous tree that grows abundantly in the dry, hilly parts of India. It is or Indian Olibanum. Since ancient times, resins have been important in the preparation of incense, medicines, cosmetics and perfumes. The Egyptians, Hindus, Persians, Israelites, Greek, Romans and the Europeans of Queen Victoria’s times greatly valued these materials. Olibanum, the resin from the *Boswellia species* has been used as incense for centuries. However, its major use today is as a fixative in perfumes, soaps, creams lotions and detergents.

In India, the gum resin exudates of *Boswellia serrata*, known in the vernacular as “Salai guggal”, has been used in the Ayurvedic system of medicine in the managements of several inflammatory conditions and as a topical anti-inflammatory agent. The major use of *Boswellia serrata* in contemporary medicine is as an anti-arthritic and anti-inflammatory pharmacological agent.

The anti-inflammatory properties of the gum resin are attributed to the presence of “boswellic acids”.

It known as Gajabhakshya in Sanskrit, implying its ingestion by elephants has been used in the Ayurvedic medicine since antiquity. The interest in this herb was aroused by the fact that such a heavy animal carried its weight on its limbs for so long, yet lived longer than humans.

This stimulated effort to find the ingredients in its diet, where *Boswellia* was found to be one. *Boswellia* has been mentioned in the ancient Indian Ayurvedic texts – the *Sushruta Samhita* and *Charak Samhita*. *Boswellia* is a tree of moderate height, which grows widely on dry hills of northwest India. In Ayurveda the oleogum resin of BSE is known as ‘*Salai Guggul*’ or ‘*Sallaki Guggul*’. It has been used in the treatment of rheumatism, nervous diseases and as a topical anti-inflammatory agent.<sup>[2]</sup>

Preparations from the gum of *Boswelliaserrata* Extract (BSE) have been used in traditional/folk medicine for treatment of inflammatory diseases. On stripping the bark, it yields gummy oleoresins, which contain oils, terpenoids and gums. Upto 16% of the resin is essential oil, the majority being  $\alpha$  thujene and p-cymene. Four pentacyclic triterpene acids are also present, with  $\beta$ -boswellic acid being the major constituent. BSE showed anti-inflammatory and antibacterial activity while the non-phenolic fraction of gum resin exhibited sedative and analgesic effects when tested in rats. Animal and in vitro studies suggest its usefulness in many inflammatory and broncho-constrictive conditions. Animal studies performed in India show that ingestion of defatted alcoholic extract of BSE decreases polymorpho-nuclear leucocyte infiltration and migration, decreased antibody synthesis and caused almost total inhibition of the classical complement pathway. Recently, it has been shown to be the inhibitor of 5-lipoxygenase and also human leucocyte elastase and consequently it has been proposed in the treatment of various inflammatory conditions.

In 1992, the active principles within the multi-component mixture of resin were identified, resulting in recognition of Boswellic acids. The most important are Acetyl 11-Keto  $\beta$ -Boswellic Acid (AKBA) and 11- Keto  $\beta$ -Boswellic Acid (KBA).<sup>[2]</sup>

**Boswellic acids were found to inhibit two pro-inflammatory enzymes, 5-lipoxygenase** (which generates inflammatory leukotrienes) and Human Leukocyte Elastase (HLE). HLE is a serine protease that initiates injury to the tissue, which in turn triggers the inflammatory process. This dual inhibitory action on the inflammatory process is unique to boswellic acids.

### **Botanical Aspects of *Boswellia* Species**

The botanical origin of *Boswellia* species has been characterized as:

- **Division:** *Spermatophyta*
- **Subdivision:** *Angiospermae*
- **Tribe:** *Rosopsida*

- **Subtribe:** *Rosidae s. lat.*
- **Overclass:** *Rutanae*
- **Class:** *Anacardiales*
- **Family:** *Burseraceae*
- **Genus:** *Boswellia*

The family of *Burseraceae* is represented in the plant kingdom with 17 genera and 600 species, widespread in all tropical regions. The species are often a predominant component of the vegetation in dry, lowland areas. Some species of the two most important genera of this family, *Commiphora* and *Boswellia*, produce resins that are of considerable commercial value as raw materials of balm, myrrh and frankincense.

## INTRODUCTION

*Boswellia* species are trees or shrubs with an outer bark often peeling in parchment like flakes, inner bark greenish, with watery aromatic resin, wood with milky latex. The leaves are imparipinnate, mostly congested at the end of the branches. Flowers are bisexual in panicles or racemes. *Boswellia* trees are found at areas from the sea level up to 1000 meters, usually in rocky slopes and gullies, often on limestone boulders, more rarely on vertical rock-faces, growing to a height of 3 up to 12 meters.

Olibanum; gum olibanum, incense or frankincense, (in German Weihrauch, Gummiresina, Kirchenharz); are the common names given to the oleo-gum resin that exudes from incisions in the bark of trees of *Boswellia* (*Burseraceae*). There are about 25 species known belonging to this genus that are widespread in India, Arabia and the northeastern coast of Africa. Since ancient times, three of these species have been considered as the “true Frankincense” producing trees. The first species grows in South Arabia. *Boswellia sacra* Flueck. is known by the Arabians as “maghrayt d’sheehaz” and the resin it produces as “lobān dhakar”.

The second one grows in Somalia, known as *Boswellia carterii* Birdw., and in native language it is called “moxor”. Recently, *Boswellia bhau-dajiana* Birdw. has been identified as identical to *B. carterii*. Generally, the resins that both species produce are called “lobān dakar” or more commonly as “beeyo” quality.

The third important olibanum is from another Somalian species, *Boswellia frereana* Birdw. Ornatively as “jagcaar”. The resin produced by this species is called “lobān majdi” or commonly “maydi”. It is the most expensive brand of olibanum on the market.

*Boswellia papyrifera* Hochst. (*B. papyrifera* Rich.) produces another olibanum quality which is known as “boido” in Somalia, Ethiopia, especially in Eritrea, in Sudan and in the other east African countries. *Boswellia neglecta* S. Moore (*B. hildebrandtii* Engl., *B. multifoliolata* Engl.) in Kenia and Ethiopia, *Boswellia rivae* Engl. (*B. boranensis* Engl) in Ethiopia, *Boswellia odorata* Hutch. And *Boswellia dalzielli* Hutch in tropical regions of Africa produce resins similar to olibanum.

Another resin producing species with a similar fragrance to frankincense is known as “Indian olibanum” or in botanical terms as *Boswellia serrata* Roxb. (syn.: *B. thurifera* Roxb., *B. thurifera* Colebr., *B. serrata* Stachh., *B. glabra* Roxb., *Canarium balsamiferum* Willd.). *B. serrata*, “salpha tree”, is found in the middle and northern parts of East India producing olibanum resin with various qualities which are commonly known as “salai guggul”.

The *Boswellia* tree contains resin channels on the bark. When the bark is incised, a white emulsion exudes and dries into globular, pear or club shaped light yellow to dark brown tears. The resin is generally harvested all through the summer and autumn after the tree has been wounded in March or April. It is supposed that a *Boswellia* tree can produce this exudate in good quality only for three subsequent years. After this period, the quality of the collected resin decreases considerably. Therefore, even in the ancient records, it has been recommended that the tree should be left to rest for some years after this harvesting period.

### Chemical History of Olibanum

Although the oil of olibanum had occupied the shelves of the 16th century pharmacies as “oleum thuris”, the first investigation on its chemical composition was performed in 1788 by **Johann Ernst Baer** at the University of Erlangen. Following his work, the first elementary analysis was carried out by **F.W. Johnston** in 1839. The constituents of the essential oil were first investigated by **J. Stenhouse** in 1840, and he identified depending on the origin of the resin fourteen monoterpenic constituents including pinene, dipentene, phellandrene and cadinene. In 1898, **A. Tschirch** and **O. Halbey** published for the first time that olibanum had an acidic constituent, boswellic acid, with a molecular formula of  $C_{32}H_{52}O_4$  but they could not suggest a structure at that time.

At the beginning of the 1930's, the olibanum resin was investigated in more detail. The study of **A. Winterstein and G. Stein in 1932** drew the attention to the resin acids, the pentacyclic triterpenoic  $\alpha$ - and  $\beta$ -amyrin like skeletons with different functional groups, which were attempted to be isolated and identified with the analytical methods possible for that time. Nevertheless, by the 1960's several of these acids such as  $\alpha$ - and  $\beta$ -boswellic acids, 11 $\alpha$ -hydroxy- $\beta$ -boswellic acid and 3-*O*-acetyl-11-hydroxy- $\beta$ -boswellic acid were identified by various derivatisation methods.

**In 1967, G. Snatzke and L. Vértés** published the structures of acetyl-11-keto- $\beta$ -boswellic acid as well as *epi*- $\alpha$ - and *epi*- $\beta$ -amyrin and their acetates,  $\alpha$ - and  $\beta$ -amyrenone and viridiflorol from the neutral fraction of olibanum, adding that it is composed of 5-9 % essential oil, 15-16 % resin acids, 25-30 % of material insoluble in ether containing the polysaccharides and 45-55 % ether soluble compounds.

**In 1978 R.S. Pardhy and S.C. Bhattacharya** identified tirucallic acids as well as  $\beta$ -boswellic acid, acetyl- $\beta$ -boswellic acid, 11-keto- $\beta$ -boswellic acid, acetyl-11-keto- $\beta$ -boswellic acid from *B. serrata* Roxb. and a diterpenoic cembrene derived alcohol, "serratol".

Studies on the isolation and identification of the boswellic acids with modern analytical techniques and on their pharmacological effects are still going on. Therefore these topics will be further discussed in the following parts of this work.

The first important and comparative study on the essential oil of olibanum of different origins was performed by **H. Obermann from Dragoco (Holzminden, Germany) in 1977**. He investigated two different commercial brands of olibanum, "Eritrea" and "Aden" by GC-MS, which corresponded to *B. carterii* and *B. serrata* resins, respectively. As a result of this investigation it was reported that not only the fragrance of these two qualities but also the composition of the constituents in the oils were different.

The "Eritrea" oil was reported to have octylacetate as the major constituent (52 %) as well as  $\alpha$ -pinene, camphene, *p*-methoxytoluol, hexyl acetate, limonene, 1,8-cineole, octanol, linalool, bornyl acetate, cembrene A, incensole, incensyl acetate and an unknown diterpenoic constituent. In contrast, "Aden" oil was found to contain  $\alpha$ -pinene as the major constituent (43%), camphene,  $\beta$ -pinene, sabinene, *o*-cymol, limonene, 1,8-cineole, *p*-cymol, campholenaldehyde, verbenone, octyl acetate and cembrenol, a diterpene alcohol with

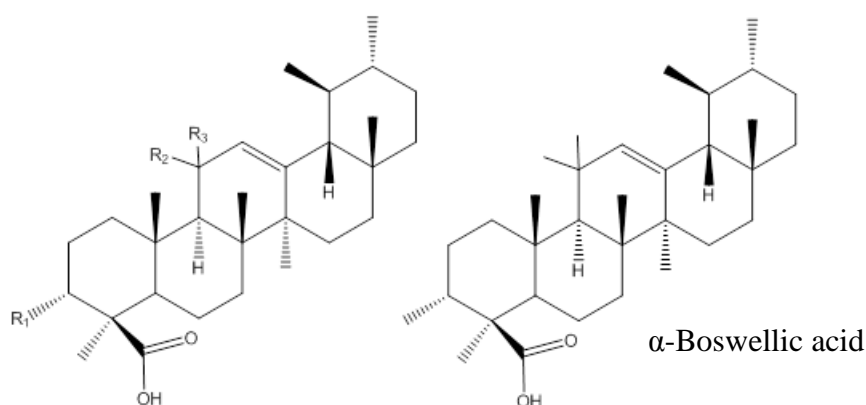
cembrene skeleton which was identified later by the same group, which was expected not be different than “serratol” described before.

In 1985 a detailed review was published by *P. Maupetit* on the “Aden” brand of olibanum. He reported 47 new constituents identified in the resinoid and in the oil of olibanum in addition to 169 formerly identified substances including the pyrolysis products. Recent studies by *Verghese on B. serrata oil* and by *A. M. Humprey et al.* comparing *B. carterii* oil with cumin, ginger, rosemary oil, were reinvestigations of known facts. These studies pointed to the difficulties in the identification of the origin of olibanum resin as well as in the determination of standard olibanum oil.

The gum resin of *Boswellia serrata* is known to contain:

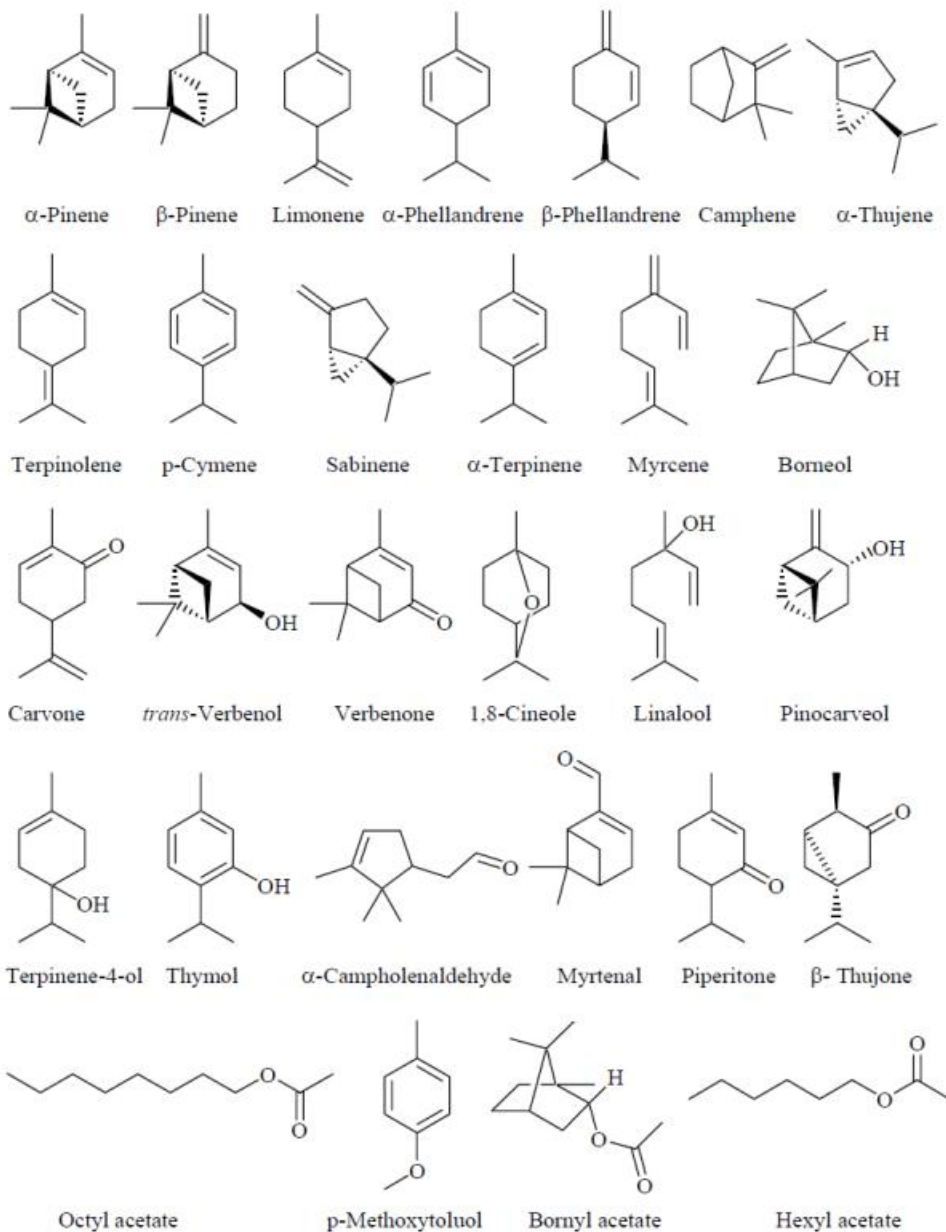
- Monoterpenes ( $\alpha$  thujene)
- Diterpenes (macrocyclic diterpenoids such as incensole, incensole oxide, inoincensole oxide, a diterpene alcohol (serrtol))
- Triterpenes (such as  $\alpha$ - and  $\beta$ -amyryns)
- Pentacyclic triterpenic acids (boswellic acids)
- Tetracyclic triterpenic acids (tirucall-8, 24-dien-21-oic acids)

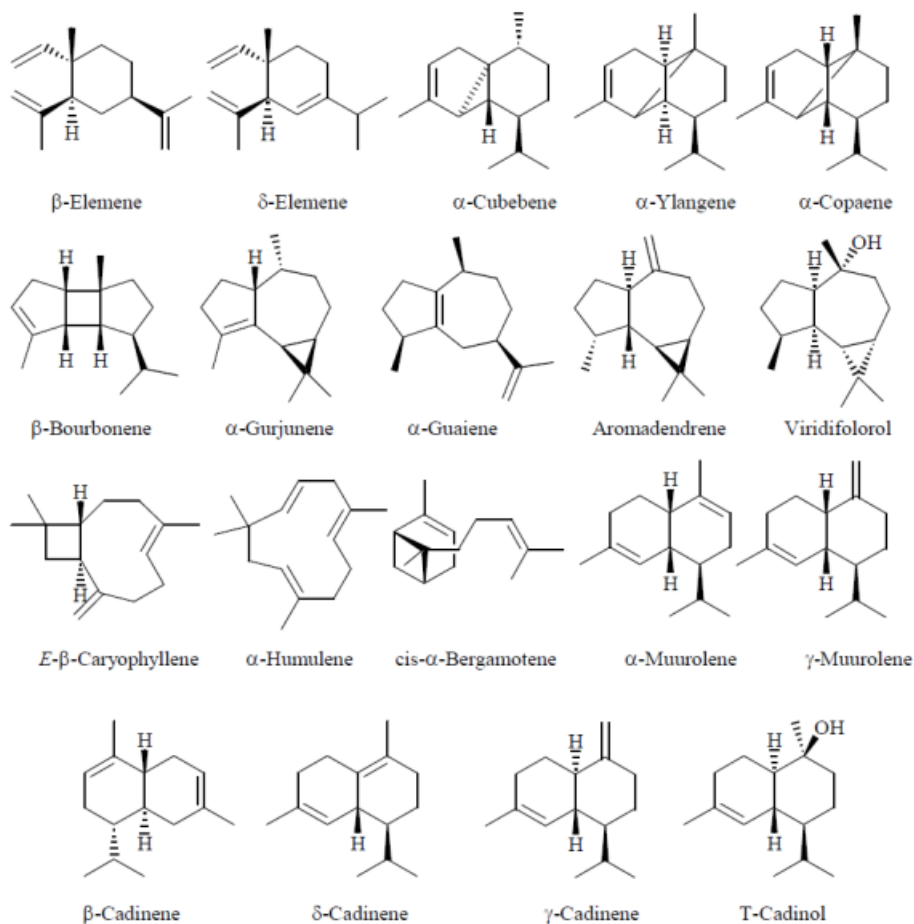
A complete list of major constituents of olibanum resin hitherto described are given in following figure,



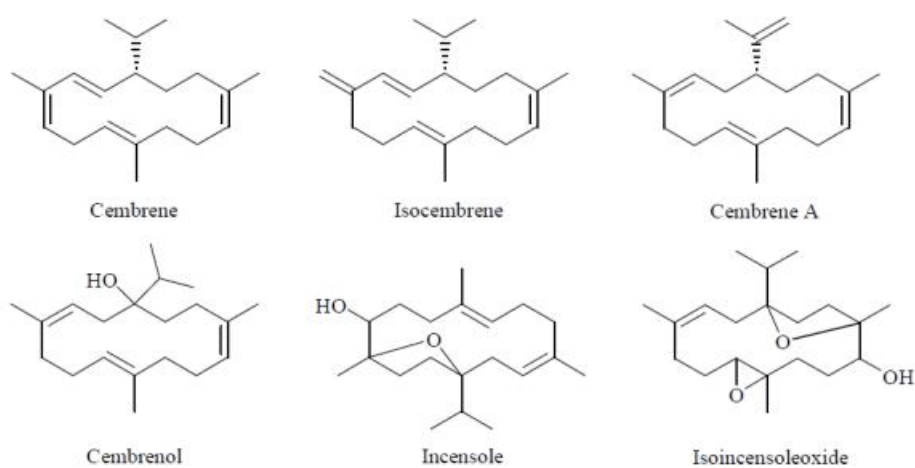
Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>
$\beta$ Boswellic acid	OH	H	H
11-hydroxy- $\beta$ -Boswellic acid	OH	OH	H

3-O-acetyl-11-hydroxy- $\beta$ -Boswellic acid	Ac	OH	H
3-O-acetyl- $\beta$ -Boswellic acid	Ac	H	H
11-keto- $\beta$ -Boswellic acid	OH	O	O
3-O-acetyl-11-keto- $\beta$ -Boswellic acid	Ac	O	O





### Sesquiterpenes



### Diterpenoids

**Table 1: Literature Survey: Phytochemical and Pharmacological.**

Sr.no	Activity	Journal	Author
1	Hepatocellular Carcinoma	Biomed Res Int. 2014;2014:294143	Khan MA1, Singh M2, Khan MS2, Najmi AK1, Ahmad S2.
2	Pediatric headache	Neurol Sci. 2014 May; 35 Suppl 1:145-8.	Dalla Libera D1, Colombo B, Pavan G, Comi G.
3	Cardioprotective and antioxidant effects	Chin J Nat Med. 2014 May;12(5):345-50.	Zaki AA1, Hashish NE2, Amer MA2, Lahloub MF2.
4	anti-inflammatory	J Nat Prod. 2014 Jun 27;77(6):1445-51	Verhoff M1, Seitz S, Paul M, Noha SM, Jauch J, Schuster D, Werz O.
5	Prevention of intestinal adenomatous polyposis	Drug Discov Ther. 2014 Feb;8(1):25-32.	Wang R1, Wang Y, Gao Z, Qu X.
6	Hypocholesteromic	J Diabetes Metab Disord. 2014 Feb 4;13(1):29.	Ahangarpour A, Heidari H1, Fatemeh RA, Pakmehr M, Shahbazian H, Ahmadi I, Mombeini Z, Mehrangiz BH.
7	Neurorecovery following injury	Brain Inj. 2013;27(12):1454-60	Moein P1, Abbasi Fard S, Asnaashari A, Baratian H, Berekatain M, Tavakoli N, Moein H.
8	Urease inhibitory activities of $\beta$ -boswellic acid	Daru. 2013 Jan 2;21(1):2	Golbabaei S1, Bazl R, Golestanian S, Nabati F, Omrany ZB, Yousefi B, Hajiaghaee R, Rezazadeh S, Amanlou M.
9	Anti-inflammatory, antinociceptive and antioxidant	Food Chem Toxicol. 2011 Oct;49(10):2594-9.	Mothana RA.
10	inhibition human monocytic (THP-1) cell activation and platelet aggregation.	J Ethnopharmacol. 2011 Sep 1;137(1):893-901	Kokkiripati PK1, Bhakshu LM, Marri S, Padmasree K, Row AT, Raghavendra AS, Tetali SD.
11	Anticancer (Boswellic acid)	Int J Cancer. 2012 May 1;130(9):2176-84.	Yadav VR1, Prasad S, Sung B, Gelovani JG, Guha S, Krishnan S, Aggarwal BB.
12	Antiinflammatory activity (preclinical)	Clin Pharmacokinet. 2011 Jun;50(6):349-69.	Abdel-Tawab M1, Werz O, Schubert-Zsilavec M.
13	Antitumor (Boswellic acid)	Food Chem Toxicol. 2011 Sep;49(9):1924-34.	Agrawal SS1, Saraswati S, Mathur R, Pandey M.
14	Effect on spatial memory retention	J Nat Med. 2011 Jul;65(3-4):519-25.	Mahmoudi A1, Hosseini-Sharifabad A, Monsef-Esfahani HR, Yazdinejad AR, Khanavi M, Roghani A, Beyer C, Sharifzadeh M.

15	Anticancer (Acetyl-11-keto- $\beta$ -boswellic acid)	Int J Cancer. 2011 Jul 1;129(1):23-33.	Park B1, Sung B, Yadav VR, Cho SG, Liu M, Aggarwal BB.
16	Antistaphylococcal and biofilm inhibitory activities (acetyl-11-keto- $\beta$ -boswellic acid)	BMC Microbiol. 2011 Mar 16;11:54.	Raja AF1, Ali F, Khan IA, Shawl AS, Arora DS, Shah BA, Taneja SC.
17	Immunological adjuvant	Int Immunopharmacol. 2011 Aug;11(8):968-75.	Gupta A1, Khajuria A, Singh J, Singh S, Suri KA, Qazi GN.
18	Anticholinesterase activity (Ethnobotanical )	Afr J Tradit Complement Altern Med. 2011;8(3):296-9.	Bakthira H1, Awadh Ali NA, Arnold N, Teichert A, Wessjohann L.
19	for inflammatory bowel disease.	World J Gastroenterol. 2010 Sep 28;16(36):4504-14.	Rahimi R, Shams-Ardekani MR, Abdollahi M.
20	Immunomodulator	Phytomedicine. 2010 Sep;17(11):862-7.	Ammon HP.
21	Asthma	Prim Care Respir J. 2010 Dec;19(4):307-14.	Clark CE1, Arnold E, Lasserson TJ, Wu T.
22	Antiplasmodial	J Ethnopharmacol. 2010 Jul 6;130(1):143-50.	Nicolas JP, De Mol P, Nikiéma JB, Frédéric M. Jansen O1, Angenot L, Tits M,
23	Memory deficit	Arch Pharm Res. 2010 Mar;33(3):463-8.	Hosseini M1, Hadjzadeh MA, Derakhshan M, Havakhah S, Rassouli FB, Rakhshandeh H, Saffarzadeh F.
24	Antioxidant activity of essential oil	Nat Prod Res. 2010;24(2):140-51.	Yang SA1, Jeon SK, Lee EJ, Shim CH, Lee IS.
25	Inhibition of vascular smooth muscle cell migration and proliferation in response to platelet-derived growth factor.	Korean J Physiol Pharmacol. 2009 Apr;13(2):107-13	Choi OB1, Park JH, Lee YJ, Lee CK, Won KJ, Kim J, Lee HM, Kim B.
26	inhibition of nitric oxide production in macrophages.	Chem Pharm Bull (Tokyo). 2009 Sep;57(9):957-64.	Yoshikawa M1, Morikawa T, Oominami H, Matsuda H.
27	inhibition of prostate tumor growth (Acetyl-11-keto-beta-boswellic acid)	Cancer Res. 2009 Jul 15;69(14):5893-900.	Pang X1, Yi Z, Zhang X, Sung B, Qu W, Lian X, Aggarwal BB, Liu M.
28	Antimalarial activity	Evid. Based Complement Alternat. Med. 2009 Dec;6(4):453-6.	Alshawsh MA1, Mothana RA, Al-Shamahy HA, Alslami SF, Lindequist U.

29	Prevention of colonic fibrosis	Eur J Clin Invest. 2008 Jun;38(6):410-20.	Latella G1, Sferra R, Vetuschi A, Zanninelli G, D'Angelo A, Catitti V, Caprilli R, Gaudio E.
30	Antiandrogen acetyl-11-keto-beta-boswellic acid	Biochem Pharmacol. 2008 Jun 1;75(11):2112-21.	Yuan HQ1, Kong F, Wang XL, Young CY, Hu XY, Lou HX.
31	The gastric ulcer protective effect (boswellic acids)	Phytomedicine. 2008 Jun;15(6-7):408-15.	Singh S1, Khajuria A, Taneja SC, Khajuria RK, Singh J, Johri RK, Qazi GN.
32	Immunomodulatory activity of biopolymeric fraction BOS 2000	Phytother Res. 2008 Mar;22(3):340-8.	Khajuria A1, Gupta A, Suden P, Singh S, Malik F, Singh J, Gupta BD, Suri KA, Srinivas VK, Ella K, Qazi GN.
33	Antiinflammatory and antiatherogenic effects acetyl-11-keto-beta-boswellic acid	Arterioscler Thromb Vasc Biol. 2008 Feb;28(2):272-7.	Cuaz-Pérolin C1, Billiet L, Baugé E, Copin C, Scott-Algara D, Genze F, Büchele B, Syrovets T, Simmet T, Rouis M.
34	anti-inflammatory - Incensole acetate novel antiinflammatory compound From boswelia resin	Mol Pharmacol. 2007 Dec;72(6):1657-64.	Moussaieff A1, Shohami E, Kashman Y, Fride E, Schmitz ML, Renner F, Fiebich BL, Munoz E, Ben-Neriah Y, Mechoulam R.
35	vaccine adjuvant (BOS 2000)	Vaccine. 2007 Jun 6;25(23):4586-94.	Khajuria A1, Gupta A, Malik F, Singh S, Singh J, Gupta BD, Suri KA, Suden P, Srinivas VK, Ella K, Qazi GN.
36	Modulation of Pgp function by boswellic acids.	Planta Med. 2006 May;72(6):507-13.	Weber CC1, Reising K, Müller WE, Schubert-Zsilavec M, Abdel-Tawab M.
37	liver damage	Y J1, Kamath JV, Asad M.	Pak J Pharm Sci. 2006 Apr;19(2):129-33.
38	Osteoarthritis of knee	Phytomedicine. 2003 Jan;10(1):3-7.	Kimmatkar N1, Thawani V, Hingorani L, Khiyani R.
39	Crohn disease	Z Gastroenterol. 2001 Jan;39(1):11-7.	Gerhardt H1, Seifert F, Buvari P, Vogelsang H, Repges R.
40	hepatitis C virus (HCV) protease inhibitory	Phytother Res. 2000 Nov;14(7):510-6.	Hussein G1, Miyashiro H, Nakamura N, Hattori M, Kakiuchi N, Shimotohno K.
41	cyclooxygenase enzyme inhibition	Nat Prod Commun. 2013 Oct;8(10):1365-6.	Ali S11, Zhang CR1, Mohamed AA1, El-Baz FK1, Hegazy AK2, Kord MA2, Nair MG3.

**Table 2: Literature Survey: Formulation.**

Sr.no	Formulation/attempt	Journal	References
1	Transdermal microemulsions <i>Boswelliacarterii</i>	Drug Deliv. 2014 Apr 14.	Mostafa DM1, Ammar NM, Basha M, Hussein RA, El Awdan S, Awad G.
2	Investigating permeability related hurdles in oral delivery of 11-keto- $\beta$ -boswellic acid.	Int J Pharm. 2014 Apr 10;464(1-2):104-10.	Bagul P1, Khomane KS1, Bansal AK2.
3	Enhanced absorption of boswellic acids by a lecithin delivery form (Phytosome) of Boswellia extract.	Fitoterapia. 2013 Jan;84:89-98.	Hüsch J1, Bohnet J, Fricker G, Skarke C, Artaria C, Appendino G, Schubert-Zsilavec M, Abdel-Tawab M.
4	Complexation with phosphatidyl choline as a strategy for absorption enhancement of boswellic acid.	Drug Deliv. 2010 Nov;17(8):587-95.	Sharma A1, Gupta NK, Dixit VK.
5	Topical Boswellic acids for treatment of photoaged skin.	Calzavara-Pinton P1, Zane C, Facchinetti E, Capezzera R, Pedretti A.	Dermatol Ther. 2010 Jan-Feb;23

**Boswellia and Boswellic acids**

The four major pentacyclic triterpenic acids present in the acidic extract of *Boswelliaserrata* gum resin.

- $\beta$ -Boswellic Acid
- Acetyl- $\beta$ -Boswellic Acid
- 11-keto- $\beta$ -Boswellic Acid
- Acetyl-11-keto- $\beta$ -Boswellic Acid

(See structures above)

Apart from this oleogum resin of boswellia also contains monoterpenes, Diterpenes and tetracyclic triterpenes (described above). These compounds are responsible for anti-inflammatory activities of resin.

**Activities of Boswellic acids**

Sr.no	Disease	Role
1	Cancer	(i) Anti-carcinogenicity in mice with ehrlic ascites carcinoma and sarcoma-180, found inhibition of tumor growth. (ii) Anti-proliferative and apoptotic effect on colon cancer. Induced anti-edema effect in glioblastoma patients.
2	Arthritis	(i) Decreases infiltration of leukocytes into knee joint and pleural cavity and inhibits the migration of polymorph nuclear leukocytes. (ii) Decreases severity of pain and disability in osteoarthritic patients.
3	Inflammation	(i) Decreases galactosamine/endotoxin induced hepatitis in mice. (ii) Decreases inflammatory features in indomethacin-induced ileitis in rats. (iii) Decreases experimental murine colitis. (iv) Inhibits the synthesis of 5-LOX products. (v) Inhibits topoisomerase, elastase and C-3 convertase enzymes.
4	Hypolipidemia	(i) Decreases cholesterol and increases HDL in rats. (ii) Induced nitric oxide production in rat macrophages. Asthma Increases stimulation of mitogen activated protein kinase MAPK and mobilization of intracellular Ca <sup>++</sup> Immunomodulatory Anti-anaphylactic and mast cell degranulation
5	Autoimmune encephalitis	(i) Inhibited ionophore stimulated release of leukotrienes from PMNL's. (ii) Decreases symptoms of AE.
6	Crohn's disease	Decreases activity index
7	Analgesic	Decreases motor activity and ptosis in rats

**Loban/Kundururu**

The oleo gum resin used in this study is also called as loban. In Ayurveda it is described as Kunduru. According to Ayurvedic Herbal Pharmacopoeia Kunduru consists of exudate of *Boswelliaserrata* Roxb. (Fam. Burseraceae), a moderate sized, deciduous tree, upto 18 m in height and upto 2.4 m in girth, commonly found in the dry forests from Punjab to West Bengal and in peninsular India.

**Synonyms**

- **Sanskrit** : sáallaki
- **Assamese** : Sallaki
- **Bengali** : Luban, Salai, Salgai
- **Gujrati** : Shaledum, Saleda, Saladi, Gugal, Saledhi
- **Hindi** : Salai, Labana
- **Kannada** : Madimar, Chilakdupa, Tallaki, Maddi

- **Kashmiri** : Kunturukkam, Samprani
- **Marathi** : Salai cha dink
- **Punjabi** : Salai Gonda
- **Tamil** : Parangi Sambrani
- **Telugu** : Parangi sambrani, Anduga, Kondagugi tamu
- **Urdu** : Kundur

### Description

#### a) Macroscopic

Drug occurs in globular, transparent, tears forming agglomerates of various shapes and sizes, brownish-yellow, upto 5 cm long, 2 cm thick, fragrant, fracture brittle; fractured surface waxy and translucent; burns readily and emanates an agreeable characteristic, balsamic resinous odor; taste, aromatic and agreeable.

### Chemical Constituents

Essential oil 8-12 %, Polysaccharides (45-60%), Higher terpenoids (25-35%).



*Boswellia serrata* tree



*Boswellia serrata* trunk



*Boswellia serrata*



branches and leaves



*Boswellia serra*



oleo gum

## Experimental Evaluation

### Procurement of Material

Oleogumresin of *Boswellia serrata* (Commonly known as Loban) was purchased from Medical Shop of crude drugs at Gulmandi, Aurangabad. 500g of material was purchased. It was available as yellowish brown to blackish masses, opaque and transparent.

### Identification test

- **General Test:** (90%) a tear of Kunduru is not altered much in form but becomes almost opaque and white; when a drop of con.  $H_2SO_4$  is added on a freshly fractured surface, it

becomes cherry red which, when washed with water changes to a white emulsion, then turn to a buff colour.

- **Fluorescence Test** - Brownish-yellow colour in day light; aqueous extract under U.V. light (366 nm) light green and in (254 nm) shows dark blue colour; alcoholic extract under U.V. light (366 nm) is colourless and in (254 nm) shows light green colour.

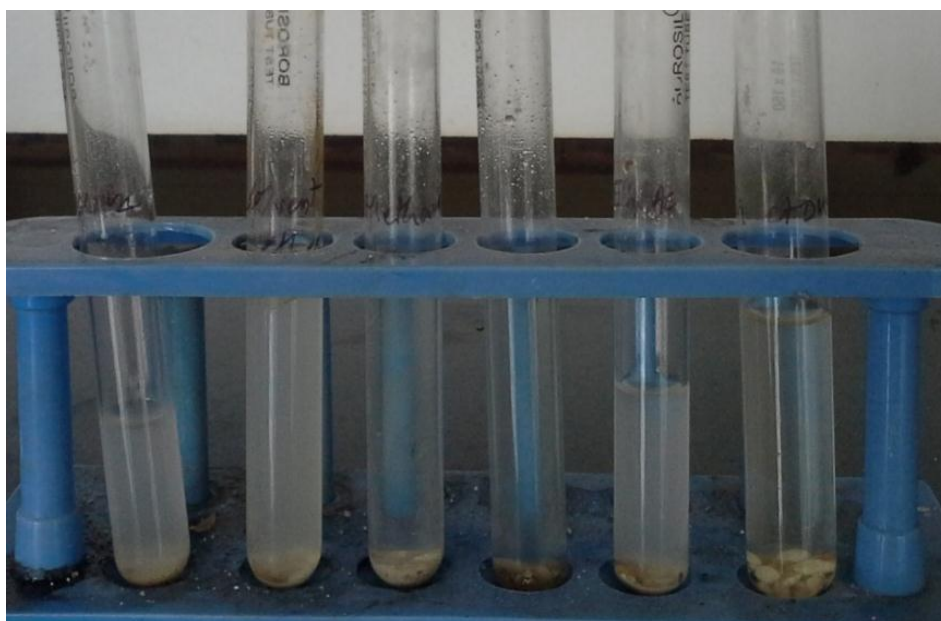
**\*Positive identification tests are given by oleogumresin**

#### Solubility studies of Oleogumresin:

For solubility study gum was crushed by mortar and pestle and its solubility was tested in different solvents. Following are the solvents used with observation. These solutions were subjected to TLC studies.

**Table 3: Solubility studies of Oleogumresin.**

Sr. no	Solvent	Appearance of solution	Residue, comments
1.	Acetone	Milky (slight cloudy)	White
2.	Ethyl acetate	Milky (slight cloudy)	Yellowish
3.	Toluene	Transperant, colourless	Yellowish
4.	Methanol	Colourless	White
5.	Ether	Milky	White yellowish
6.	n Hexane	Slight cloudy	Brown
7.	Ethanol	Clear yellow transperant	Amorphus white
8.	Water	Yellowish turbid solution	Waxy lumps of white colour



### Thin Layer Chromatography(TLC)

- **Sample Preparation**

In case of dried extract dissolve a small quantity of the sample in the least polar solvent in which it is soluble. Alternatively sample can also be prepared by extracting the small amount of crude drug material (say 1g) with solvent.

- **Sample Application**

Sample thus prepared is applied in very small amount (in microlitres) as spot or as lane on TLC plate just 1cm-2cm above the base of it. This is called as solute front.

- **Development of TLC Plate**

After the TLC plate is spotted it has to be transferred to the development tank. This tank holds the mobile phase. The solute front is not allowed to sink in to the solvent system otherwise the compound will diffuse in to the solvent system. Large variety of development tanks are commercially available, the cheapest way is to use a beaker with a watch glass as lid. To saturate the inside atmosphere of the tank, the tank can be lined with filter paper.

- **Visualization of TLC Plate/ Detection of Spots**

It is easy to visualize coloured compounds but all compounds are not coloured. There are numerous reagents available to visualise TLC plates. Even when the material being analyzed is coloured it is necessary to treat the TLC plate to visualize any no-coloured spots that may be present in the sample. The two most useful means of analysis are ultraviolet-light and iodine vapour. The TLC plate can be dipped into a stock solution of the reagent or the plate can be sprayed with a diffuser. Commonly used reagents are anisaldehyde sulphuric acid reagent, vanillin sulphuric acid reagent, antimony trichloride reagent. After the plate has been sprayed, it is heated for 10 minutes at 110°C. The compounds produce spots visible under ultraviolet light 360 nm and 254nm.

- **Stationary phase:** Pre coated silica gel plate with fluorescent indicator (F<sub>254</sub>) from Merck PSGF<sub>254</sub>

- **Mobile Phase:** Toluene: Ethyl Acetate: Methanol (8:2:1)

- **Detection**

1. UV 254nm
2. Anisaldehyde sulphuric acid (ASA) (Heat treatment is required after TLC plate is sprayed with ASA).

### Extraction Studies

Crude drug material is subjected for extraction studies by employing different methods of extraction commonly available in laboratory.

Following methods were tested

1. Maceration
2. Hot continuous extraction (Soxhlet extraction)
3. Steam distillation and reflux
4. Direct heating (Decoction)

**Maceration:** In this method material 250g of oleogumresin is subjected for maceration for 20 hrs in ethanol (500ml) with continuous shaking at 100 rpm on shaker. This process is repeated successively for three times.

**Hot continuous Extraction:** This is also called as hot continuous percolation or soxhlet extraction. Crude drug material is placed in body of extractor. Heating is applied.

**Steam distillation or reflux:** In this method oleogumresin is placed in RBF fitted with condenser and heat is applied to RBF.

**Direct Heating:** In this method oleogum resin is placed in a glass container and direct heat is applied to container.

### Isolation of Boswellic Acid by Column Chromatography

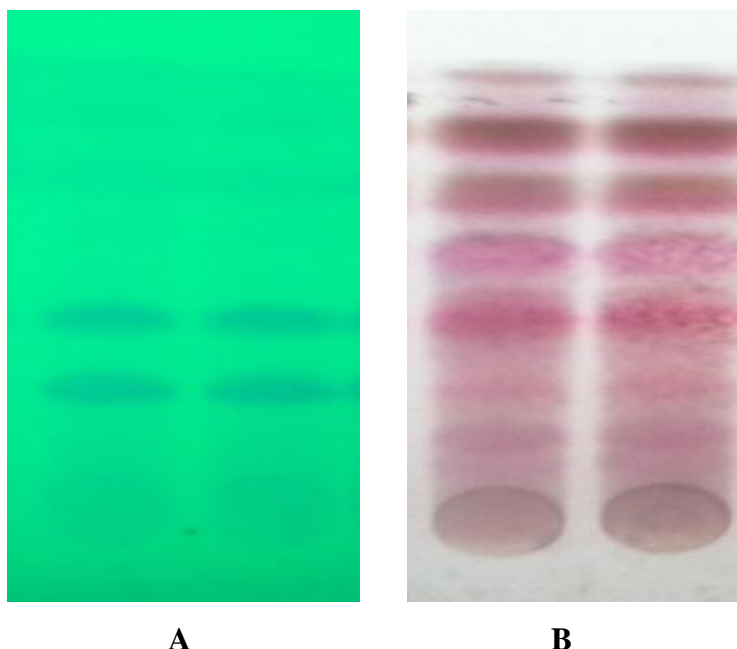
For running of column optimization of solvent system is carried out by Thin Layer Chromatography.

- **Thin Layer Chromatography**

Thin layer chromatography is carried out under following conditions

- **Mobile Phase:** Toluene: Ethyl acetate: n heptane: Formic acid (8:2:1:0.3)
- **Stationary phase:** Precoated silica gel plate with fluorescent indicator (F254) from Merck PSGF<sub>254</sub>
- **Detection**
  - A. UV 254 nm
  - B. Anisaldehyde sulphuric acid (ASA)

Heat treatment is required after TLC plate is sprayed with ASA



**Photograph 4.1: TLC of Boswellia extract**

**(Toluene: Ethyl acetate: n heptane: Formic acid)**

### **Column Chromatography**

#### **Procedure**

In column chromatography, the mobile phase is again a solvent, and the stationary phase is a finely divided solid, such as silica gel or alumina. Chromatography columns vary in sizes. There is an element of trial and error involved in selecting a suitable solvent and adsorbent for the separation of the constituents of a particular mixture. A small volume of the sample whose constituents are to be separated is placed on top of the column. The choice of the eluting solvent should ensure that the sample is soluble. However, if the sample was too soluble the mobile phase (solvent) would move the solutes too quickly, resulting in the non-separation of the different constituents. Usually, one should start with a less polar solvent to remove the less polar compounds, and then slowly increase the polarity of the solvent to remove the more polar compounds.



**Photograph 4.2: Steps in Column Chromatography.**

### Column Chromatography 1

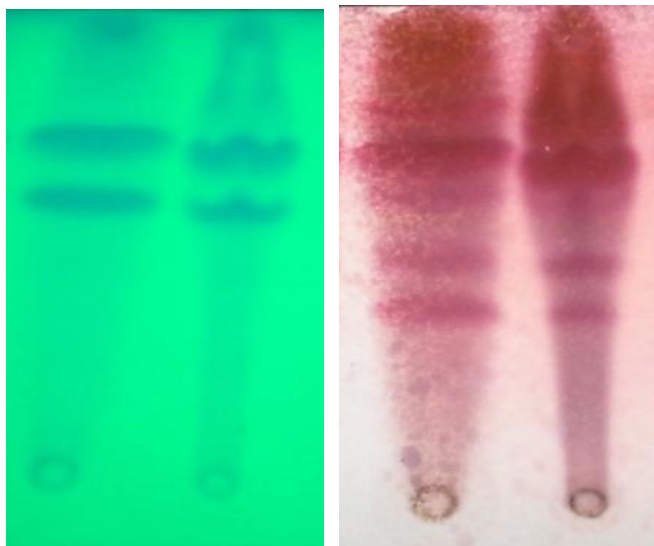
- **Stationary Phase:** Silica 230-330 mesh size.
- **Mobile phase:** Toluene: Ethyl acetate: n heptane (8:2:1)
- **Column:** Glass column
- **Column diameter** (internal diameter of column):1.5cm
- **Height of column:** 10.5cm
- **Height of extract loaded:** 1.5cm  
(Height of column refers to height at which silica, the stationary phase is filled)
- **Number of fractions collected:** 32 fractions of 20ml each.  
Fraction 4, 5 shows single spot corresponds to Acetyl keto  $\beta$  boswellic acid  
Fraction 8-9 yields single spot corresponds to  $\alpha$  and  $\beta$  boswellic acid

### Column Chromatography 2

This column was run on Toluene: Ethyl acetate: Methanol (8:2:1). This mobile phase was optimized with TLC.

- **Thin Layer Chromatography**  
Thin layer chromatography is carried out under following conditions
- **Mobile Phase:** Toluene: Ethyl acetate: methanol (8:2:1)
- **Stationary phase:** Precoated silica gel plate with fluorescent indicator (F254) from Merck PSGF<sub>254</sub>
- **Detection**
  - Visible (Vis.)-without any chemical treatment-
  - Anisaldehyde sulphuric acid (ASA)-Plate 2
  - UV 254 nm

- Heat treatment is required after TLC plate is sprayed with ASA



**Photograph 4.3: TLC of Boswellia extract  
(Toluene: Ethyl acetate: methanol)**

### Column Chromatography

- **Stationary Phase:** Silica 230-330 mesh size.
- **Mobile phase:** Toluene: Ethyl acetate: n heptane: Formic acid (8:2:1:0.3)
- **Column:** Glass column
- **Column diameter** (internal diameter of column):1.5cm
- **Height of column:** 10.5cm
- **Height of extract loaded:** 1.5cm

(Height of column refers to height at which silica, the stationary phase is filled)

- **Number of fractions collected:** 32 fractions of 10ml each.

Fraction 4, 5 shows single spot corresponds to Acetyl keto  $\beta$  boswellic acid

Fraction 8-9 yields single spot corresponds to  $\alpha$  and  $\beta$  boswellic acid

### Isolation of acid fraction

**Following general procedure is employed for isolation of boswellic acid from oleo gum resin**

- (a) Crushing the lumps of the gum resin of *Boswellia serrata* and extracting the crushed lumps with ethanol.
- (b) Removing insoluble material from above extract;
- (c) Concentrating the extract till a reddish brown syrupy mass is obtained;

- (d) Basifying the syrupy mass with an aqueous solution of an alkali to provide a solution having a pH in the range of 9 to 10.
- (e) Extracting the solution with a solvents to provide an aqueous layer, and acidifying the aqueous layer with mineral acid to a pH in the range of 3-5 to provide a precipitate comprising boswellic acids;
- (f) Washing the precipitate with water to provide said fraction being neutral to litmus;
- (g) Separating individual boswellic acids from said fraction.

### **Checking effect of alkali**

In the procedure mentioned above step (d) which indicates use alkali for basification. The aim of the present study is to check effect of different concentrations of alkali (NaOH and KOH) on separation of acid fraction.

### **Koh**

#### **Acid separation by using 2% koh**

10g of sample dissolved in ethanol. This solution was treated with 2% KOH solution (pH=10) and acidified with Conc. HCl.

#### **Acid separation by using 5% koh**

10g of sample dissolved in ethanol. This solution was treated with 5% KOH (pH 10.5) and acidified with conc. HCl. If no Precipitate is formed and extract remain as such without getting dissolved as viscous orange mass. This orange mass was separated and dissolved in ethanol and this solution was treated with Conc. HCl. Precipitate is formed which is separated and dissolved in ethyl acetate. It gave a soluble portion and insoluble precipitate. Insoluble precipitate was separated and dissolved in ethanol.

#### **Acid separation by using 10% koh**

10g of sample dissolved in ethanol. This solution was treated with 10% KOH (pH 10.5) and acidified with conc. HCl. If no Precipitate is formed and extract remain as such without getting dissolved as viscous orange mass. This orange mass was separated and dissolved in ethanol and this solution was treated with Conc. HCl. Precipitate is formed which is separated and dissolved in ethyl acetate. It gave a soluble portion and insoluble precipitate. Insoluble precipitate was separated and dissolved in ethanol.

**Acid separation by using 20% koh**

10g of sample dissolved in ethanol. This solution was treated with 10% KOH (pH 10.5) and acidified with conc. HCl. If no Precipitate is formed and extract remain as such without getting dissolved as viscous orange mass. This orange mass was very less which separated and dissolved in ethanol and this solution was treated with Conc. HCl. No precipitate was formed.

**Naoh**

10g of sample dissolved in ethanol. This solution was treated with 2% NaOH (pH 10.5) and acidified with conc. HCl. If no Precipitate is formed and extract remain as such without getting dissolved as viscous orange mass. This orange mass was separated and dissolved in ethanol and this solution was treated with Conc. HCl. Precipitate is formed which is separated and dissolved in ethanol.

**5% Naoh**

10g of sample dissolved in ethanol. This solution was treated with 5% NaOH and acidified with conc. HCl. The extract remain as such without getting dissolved as viscous orange mass. This orange mass was separated and dissolved in ethanol and this solution was treated with Conc. HCl. Precipitate is formed which is separated and dissolved in ethanol.

**10% Naoh**

10g of sample dissolved in ethanol. This solution was treated with 10% NaOH and acidified with conc. HCl. The extract remain as such without getting dissolved as viscous orange mass. This orange mass was separated and dissolved in ethanol and this solution was treated with Conc. HCl. Precipitate is formed which is separated and dissolved in ethanol.

**20% Naoh**

10g of sample dissolved in ethanol. This solution was treated with 10% NaOH and acidified with conc. HCl. The extract remained as such without getting dissolved as viscous orange mass. This orange mass was separated and dissolved in ethanol and this solution was treated with Conc. HCl. Precipitate is formed which is separated and dissolved in ethanol

**Checking effect of acid**

In this parameter same above procedures were repeated for NaOH and KOH. Conc. H<sub>2</sub>SO<sub>4</sub> was used instead of Conc. HCl.

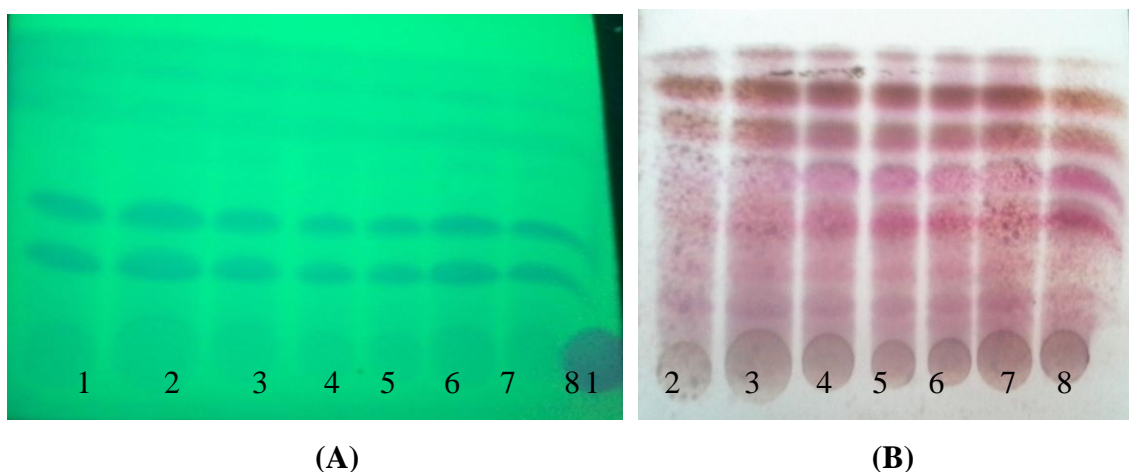
### Effect of strength of acid

Here same procedures were repeated dilute acid was used instead of concentrated acid.

## OBSERVATION AND RESULTS

### Selection of suitable solvent

As mentioned above in experimental work done, oleo gum resin was subjected for solubility studies in different solvents. In this 10 commonly available solvents were used for extraction including water. All of them showed same TLC pattern except water. This indicates that any solvent can be used for extraction of oleo gum resin except water (Use of any solvent here is recommended if the aim of extraction is isolation of Boswellic acid). Here one important point of cost can be considered with respect to solvent which will influence choice of solvent for extraction.



**Photograph 5.1: TLC fingerprint for effect of solvents**

1. Ac- acetone
2. Etac-ethyl acetate
3. Tol- Toluene
4. Met-Methanol
5. Eth-Ethanol
6. Hex-Hexane
7. Chl- Chloroform
8. Wat- water

### Details of TLC

- **Thin Layer Chromatography**

Thin layer chromatography is carried out under following conditions

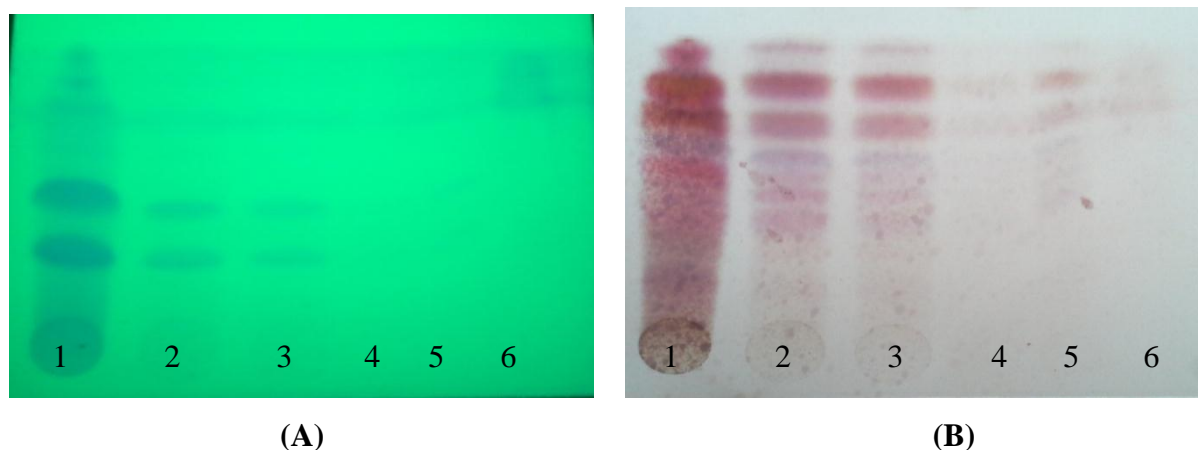
- **Mobile Phase:** Toluene: Ethyl acetate: n heptane: Formic acid (8:2:1:0.3)
- **Stationary phase:** Precoated silica gel plate with fluorescent indicator (F<sub>254</sub>) from Merck PSGF254
- **Detection**
  - (A)-UV 254 nm
  - (B)- Anisaldehyde sulphuric acid (ASA)
- Heat treatment is required after TLC plate is sprayed with ASA

### Selection of extraction method of oleogum resin and completeness of extraction

As mentioned in experimental conditions above different extraction methods were used for oleogum resin. Oleogum resin is mixture of polysaccharides gum, volatile oil and resin. Resin portion contains boswellic acids.

In case of soxhlet extraction and other methods of extraction employing heat like decoction problem of clogging and melting of material arises, which hampers process of extraction.

In case of maceration no heat is employed, but the material has to be extracted multiple times with fresh solvent each time. In this method solvent requirement increases. Each time the extract was checked for presence of boswellic acids by TLC to ensure complete extraction. It took 4-5 times repeated extraction for complete extraction of Boswellic acids by maceration. After extraction concentration of material was carried out. During concentration much of the volatile oil evaporates.



Photograph 5.2: TLC fingerprint for completeness of extraction

### Thin Layer Chromatography

Thin layer chromatography is carried out under following conditions

- **Mobile Phase:** Toluene: Ethyl acetate: n heptane: Formic acid (8:2:1:0.3)
- **Stationary phase:** Precoated silica gel plate with fluorescent indicator (F<sub>254</sub>) from Merck PSGF254
- **Detection**
  - (A)-UV 254 nm
  - (B)- Anisaldehyde sulphuric acid (ASA)

Heat treatment is required after TLC plate is sprayed with ASA

### Isolation of total acids (boswellic acid)

Literature survey reveals that anti-inflammatory activities associated with this resin are completely restricted to presence of Boswellic acids. So focus in experimental work done is placed on isolation of acid fraction of oleo gum resin.

As mentioned in the procedures above acid fraction was obtained as white precipitate. This white precipitate was separated, dried and weighed.

Amount of acid fraction obtained was determined on weight basis.

In the experimental work done isolation of acid fraction was carried out by procedures mentioned in section above. This is common procedure which uses treatment of resin with alkali to convert acid into its salt and then precipitating salt of acid by using mineral acid.

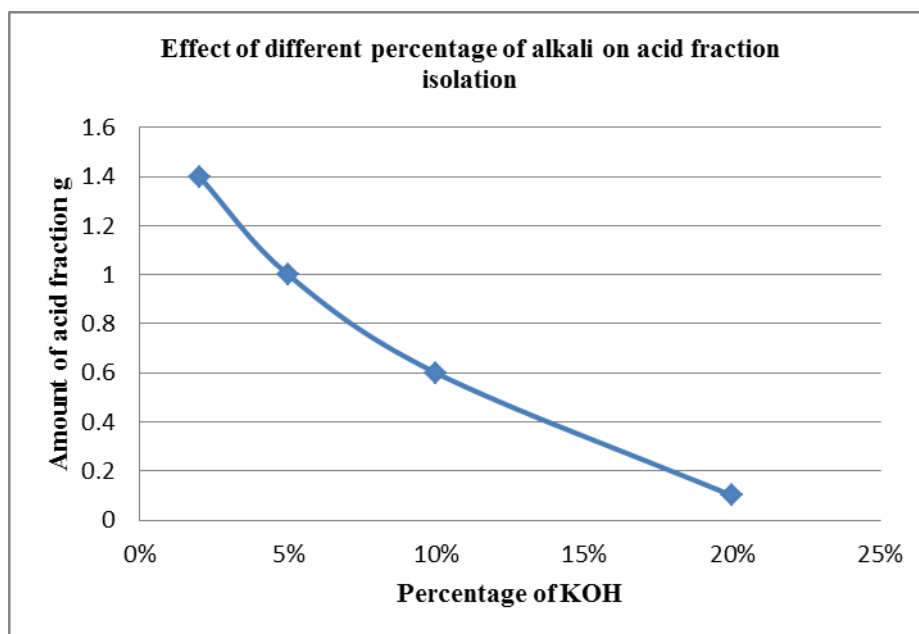
Studies were carried out to check variations in amount of acid portion obtained when these parameters were altered.

### Effect of koh

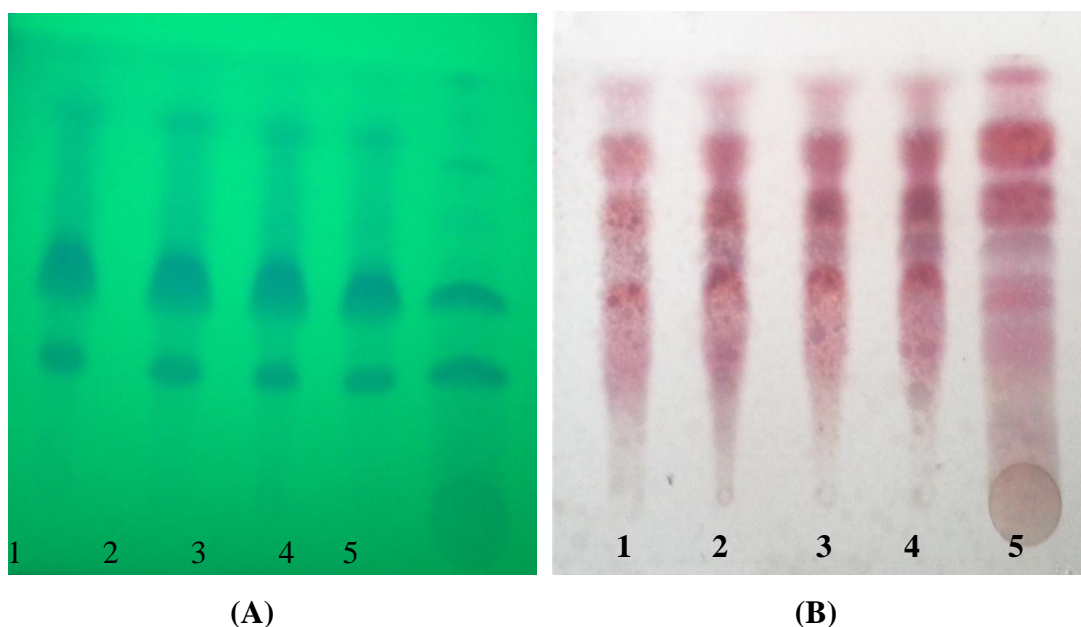
Following are observations when KOH is used as basifying agent.

**Table 5.1: Effect of percentage of KOH as basifying agent on acid fraction.**

Sr.no	Percentage of KOH	Amount of acid
1	2% KOH	1.4g
2	5% KOH	1.0g
3	10% KOH	0.6g
4	20% KOH	0.1g



Graph 5.1: Effect of different percentage of alkali on acid fraction isolation.



Photograph 5.3: TLC fingerprint for effect of KOH

1. 2% KOH
2. 5 % KOH
3. 10% KOH
4. 20 % KOH
5. Whole extract

### Thin Layer Chromatography

Thin layer chromatography is carried out under following conditions

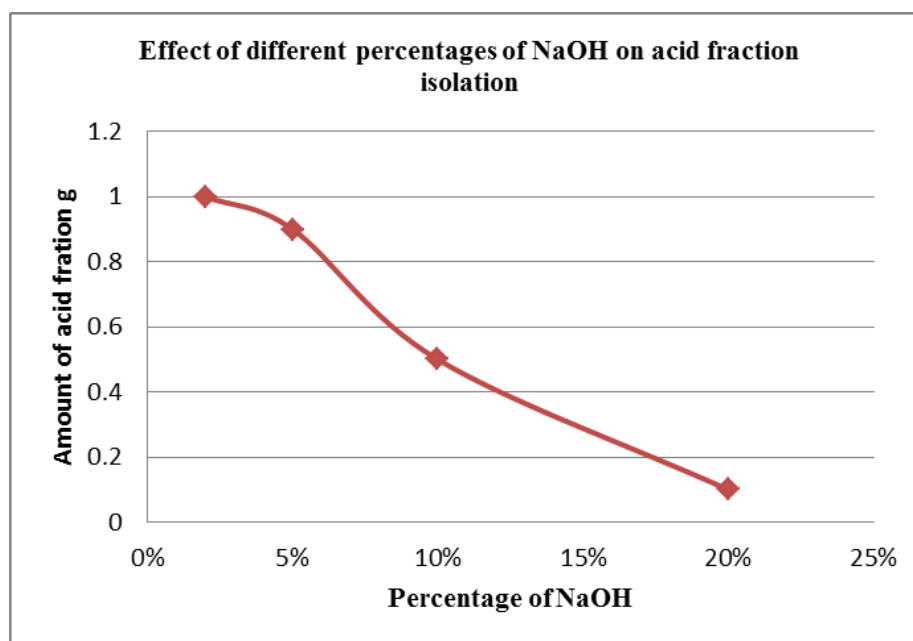
- **Mobile Phase:** Toluene: Ethyl acetate: n heptane: Formic acid (8:2:1:0.3)
- **Stationary phase:** Precoated silica gel plate with fluorescent indicator (F<sub>254</sub>) from Merck PSGF<sub>254</sub>
- **Detection:**
  - (A)-UV 254 nm
  - (B)- Anisaldehyde sulphuric acid (ASA)
- Heat treatment is required after TLC plate is sprayed with ASA

### Effect of Naoh

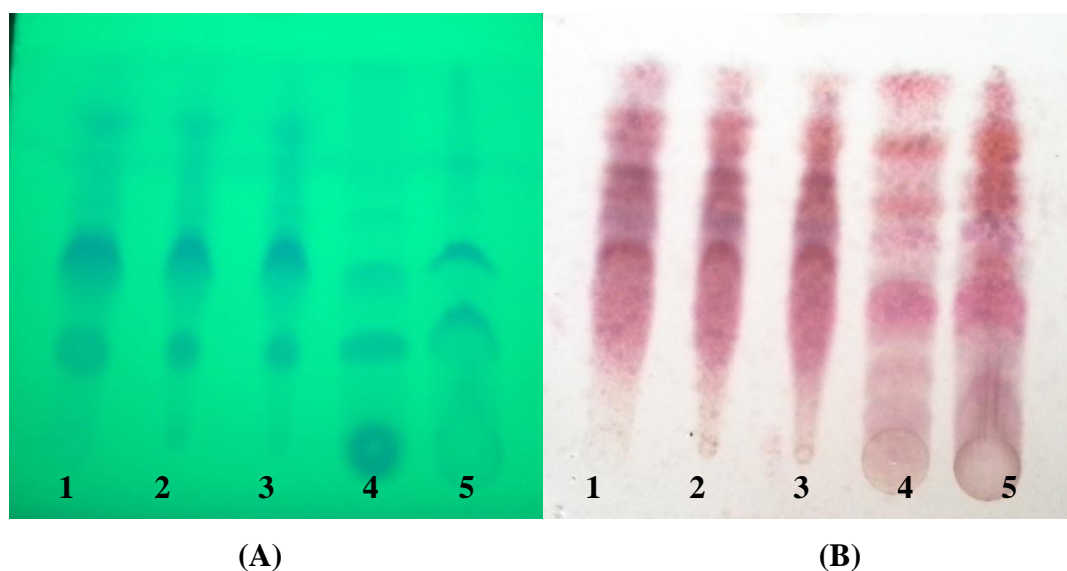
Following are observations when NaOH is used as basifying agent.

**Table 4: Effect of percentage of NaOH as basifying agent on acid fraction.**

Sr. no	Percentage of NaOH	Amount of acid
1	2% NaOH	1.0g
2	5% NaOH	0.9g
3	10% NaOH	0.5g
4	20% NaOH	0.1g



**Graph 5.2: Effect of different percentages of NaOH on acid fraction isolation.**



TLC fingerprint for effect of NaOH

1. 2% NaOH
2. 5 % NaOH
3. 10% NaOH
4. 20 % NaOH
5. Whole extract

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